

To practice these skills, complete this table

<u>Symbol</u>	<u>Atomic number</u>	<u>Mass number</u>	<u># of protons</u>	<u># of neutrons</u>	<u># of electrons</u>
_____	27	59	_____	_____	27
Au ³⁺	_____	_____	_____	118	_____
_____	_____	32	15	_____	18
_____	_____	81	35	_____	36
K ¹⁺	_____	_____	_____	20	_____
_____	36	84	_____	_____	36
S ²⁻	_____	_____	_____	16	_____
_____	29	63	_____	_____	29
I ¹⁻	_____	_____	_____	74	_____
_____	_____	18	8	_____	10
_____	50	119	_____	_____	50